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⑤④ A polyurethane foam composition suitable for backing a substrate and a substrate containing the polyurethane foam composition.

⑤⑦ This invention is directed to a polyurethane foam composition which is suitable as a backing for a substrate and to a substrate containing the polyurethane foam composition. The foam composition is the reaction product of a frothed polyurethane forming mixture comprising a relatively high molecular weight polyether polyol, a relatively low molecular weight polyol, an organic polyisocyanate or polyisothiocyanate, an inorganic filler, a catalyst for urethane formation, and a silicone surfactant. The silicone surfactant is a poly(dimethylsiloxane) polyoxyalkylene branch copolymer having an average molecular weight of less than 30,000 wherein at least 60 percent by weight of the polyoxyalkylene groups are oxyethylene groups, the dimethylsiloxane content of the copolymer being from 15 to 40 percent by weight, at least 40 percent of the weight of the copolymer is derived from the oxyethylene groups. The silicone surfactant is present in the mixture in a quantity from 0.02 to less than 0.5 parts by weight per 100 parts by weight of the combined weights of the relatively high molecular weight polyether polyol and the relatively low molecular weight polyol.

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A POLYURETHANE FOAM COMPOSITION SUITABLE
FOR BACKING A SUBSTRATE AND A SUBSTRATE CONTAINING
THE POLYURETHANE FOAM COMPOSITION

The present invention pertains to a polyurethane foam composition which is the reaction product of a frothed polyurethane forming mixture and to the resultant polyurethane foam backed substrates.

5 Substrates, particularly carpet substrates,
have been backed with polyurethane foam as described in
U.S. Patents 3,755,212, Dunlap and Parrish; 3,821,130,
Barron and Dunlap; and 3,862,879, Barron and Dunlap
wherein the frothed polyurethane forming composition
10 does not contain a silicone surfactant and in U.S.
Patents 3,706,681, Bachura; 3,849,156, Marlin et al.;
3,836,560, 3,947,386, 3,957,842, 4,022,722 and
4,022,941, Prokai et al. wherein the frothed poly-
urethane forming compositions contain a silicone
15 surfactant.

The polyurethane foam compositions of the
present invention containing polyoxyalkylene siloxane
branch copolymers have fine cells and excellent
stability during processing. Furthermore, the compo-
20 sitions of the present invention incorporate low

0048986

amounts of the silicone surfactant(s) (stabilizer), and have improved esthetics and cost over the same polyurethane foams without the silicone surfactant (non-stabilized) or containing linear siloxane surfac-

5 tants. It is known that penetration weight (weight of the polyurethane composition absorbed by a substrate) is dependent upon several factors, including type of substrate, component reactivity, temperature and compo-

10 nent viscosity. Penetration weight on the commercially available carpet described in this invention had at least about 25 percent reduction compared to non-

15 -stabilized formulations. The reduced penetration and improved stability has resulted, on the average, in a 20 percent improvement in gauge (thickness of foam)

20 over non-stabilized formulations. Cost to the customer has been effectively reduced by the process of this invention because a product (foam backed substrate) is obtained that has the same thickness employing less components by weight. In other terms, the same weight of froth is applied to the back of the substrate and a desirably thicker foam is obtained.

It is well known in the art that the frothing or mechanical whipping operation of the polyurethane forming composition is performed in high shear equip-

25 ment such as a Hobart mixer, Firestone mixer, Oakes mixer and similar known equipment. The air or other inert gas should be introduced into the polyurethane forming mixture and whipped under pressure in order to obtain a workable froth. The mechanical whipping

30 operation is conducted such that the pressure at the mixer head outlet is in the range from about 85 to about 200 psig (1.48 MPa), preferably from 85 to 150 psig (0.69 to 1.14 MPa).

0048986

In those processes where there is no silicone surfactant present, the substrates absorb an inordinate amount of polyurethane forming components thereby resulting in too much absorption beyond what is necessary for tuftbinding. A loss of gauge and a destabilization line at the substrate/foam interface occurs, wherein the cell structure becomes coarse and unappealing. The loss of gauge and destabilization line are esthetically unappealing to the buyer and results in higher costs to the vendor. The solution to this without a silicone is a high catalyst level or quick reaction time, which severely limits processing latitude.

In those processes wherein a silicone surfactant is employed, large quantities of the silicone surfactant are required in order to obtain fine cells or else stability of the polyurethane forming froth is not sufficient to give a fine celled foam or prevent an inordinate amount of penetration (absorption). In the present invention, a large quantity of surfactant (>0.5 part) limits the processing of the froth because froth viscosity is too high to be adequately processed onto the back of a substrate such as a carpet and the pressure in the mixer head may exceed the working limit of 200 psig (1.48 MPa). Also, in the present invention, the improvement in penetration and gauge is no longer measurable from greater than 0.3 parts surfactant. There is no real advantage, therefore, to using surfactant levels above 0.5 parts from a processing or cost standpoint.

0048986

The present invention is directed to a polyurethane foam composition which is the reaction product of a frothed polyurethane forming mixture comprising:

- 5 (A) a polyethylene polyol having an average of from 2 to 8 hydroxyl groups per molecule and an average hydroxyl equivalent weight of from 100 to 3000;
- 10 (B) a polyol having an average of from 2 to 8 hydroxyl groups per molecule and an average hydroxyl equivalent weight of from 31 to 230, component (B) being optional when component (A) contains an end cap of at least two moles of ethylene oxide per active hydrogen atom or an internal block of at least three moles of
- 15 ethylene oxide residues per active hydrogen atom, when components (A) and (B) are employed together, they are employed in proportions providing from 50 to 95 parts by weight of component (A) and from 50 to 5 parts by
- 20 weight of component (B) based upon 100 parts by weight of the total quantity of components (A) and (B), the hydroxyl equivalent weight of component (A) is at least 1.5 times greater than the hydroxyl equivalent weight
- 25 of component (B), the mixture of components (A) and (B) or component (A) when employed alone has a Brookfield viscosity employing a number 2 spindle at 12 rpm of at least 100 centipoise (0.1 Pascal-second) at 25°C;
- 30 (C) an organic polyisocyanate or polyisothiocyanate having an average of from 2 to 4 NCX groups per molecule, each X being independently oxygen or sulfur, components (A), (B)

0048986

and (C) being employed in quantities so as to provide an NCX/ ϕ^4 ratio of from 0.85:1 to 2:1;

- (D) at least one inorganic filler;
- 5 (E) at least one catalyst for reacting components (A) and (B) with component (C), component (E) being employed in a quantity such that an unfrothed mixture of components (A), (B),
10 (C), (E) and (F) obtains a Brookfield viscosity employing a number 4 spindle at 30 rpm greater than 10,000 centipoise (10 Pascal-seconds) at 25°C in less than 40 minutes; and
15 (F) a silicone surfactant, characterized in that component (F) is a poly(dimethylsiloxane) polyoxyalkylene branch copolymer having an average molecular weight of less than 30,000, preferably less than 10,000, wherein at least
20 60 percent by weight, preferably 100 percent by weight, of the polyoxyalkylene groups are oxyethylene groups, the dimethylsiloxane content of the copolymer being from 15 to 40 percent by weight, and at least 40 percent of the weight of the copolymer is derived from the oxyethylene groups, component (F) being
25 present in a quantity from 0.02 to less than 0.5 parts by weight per 100 parts by weight of the combined weights of components (A) and (B). Preferably, the foam has a cell count of greater than 16 per 0.0025 square inches
30 (1.6 mm²) at a position approximately midway between the carpet-foam interface and the outer edge of the foam. Preferably, the resultant cured foam has a density of less than 13 lb/ft³ (208 Kg/m³).

0048986

Suitable relatively high molecular weight polyether polyols which can be employed in the present invention include adducts of one or more compounds containing 2 to 8 hydroxyl groups per molecule and one
5 or more hydrocarbylene or halogen substituted hydrocarbylene oxides.

Suitable hydroxyl containing compounds include those having from 2 to 20, preferably from 2 to 6 carbon atoms such as, for example, ethylene glycol,
10 diethylene glycol, propylene glycol, 1,2-butane diol, 1,3-butane diol, 1,4-butane diol, glycerine, trimethylol propane, p,p'-isopropylidene diphenol, and mixtures thereof.

Suitable hydrocarbylene or halogen substituted hydrocarbylene oxides which can be employed to
15 prepare polyether polyols include those having from 2 to 12, preferably from 2 to 4 carbon atoms such as, for example, ethylene oxide, propylene oxide, 1,2-butylene oxide, 2,3-butylene oxide, styrene oxide, epichlorohydrin, epibromohydrin, 3-methyl-1,2-butylene oxide,
20 3,3-dimethyl-1,2-butylene oxide, and mixtures thereof.

Other relatively high molecular weight polyols which can be employed herein include polymer-containing polyols such as, for example, those disclosed in U.S. Patents RE 29,118 (Stamberger), RE
25 28,715 (Stamberger), RE 29,014 (Pizzini et al) and 3,869,413 (Blankenship et al).

Suitable low molecular weight polyols include, for example, ethylene glycol, propylene
30 glycol, 1,3-propane diol, 1,4-butane diol, dipropylene

0048986

glycol, diethylene glycol, triethylene glycol, tetraethylene glycol, tripropylene glycol, tetrapropylene glycol, heptapropylene glycol, and mixtures thereof.

Also suitable as low molecular weight polyols
5 are adducts of initiator compounds having from 2 to 8 active hydrogen atoms per molecule and a hydrocarbylene oxide as hereinbefore described so long as the resultant adduct has the desired equivalent weight for the low equivalent weight polyol.

10 Particularly suitable initiator compounds include, for example, water, ethylene glycol, propylene glycol, glycerine, trimethylol propane, aniline, ammonia, ethylene diamine, diethylenetriamine, amino-ethylethanolamine, pentaerythritol, glucose, fructose,
15 sucrose, sorbitol, and mixtures thereof.

When adducts of initiator compounds having more than 8 active hydrogen atoms per molecule are employed, they must be employed in admixture with polyols having less than 8 hydroxyl groups per molecule
20 in quantities so as to result in an average of from 2 to 8 hydroxyl groups per molecule.

When the viscosity of a mixture of polyol components (A) and (B) or the viscosity of polyol (A) when employed alone is below 100 centipoise (0.1 Pascal-
25 -second) at 25°C, the resultant filled, foam-forming system is too fluid to froth adequately and does not result in a fine celled, low density foam.

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Suitable organic polyisocyanates include, for example, 2,4-toluenediisocyanate, 2,6-toluenediisocyanate, xylylenediisocyanate, p,p'-diphenylmethane-diisocyanate, p-phenylenediisocyanate, naphthalenediisocyanate, dianisodine diisocyanate, polymethylene polyphenylisocyanate, hexamethylene diisocyanate, and mixtures thereof.

Also suitable are the isocyanate terminated prepolymers prepared from one or more of the above mentioned polyisocyanates and one or more of the above mentioned polyols.

Suitable also are the trimerized diisocyanates and crude diisocyanates.

Also suitable are the corresponding polyisothiocyanates, including isothiocyanate terminated prepolymers.

Suitable catalysts which can be employed include, for example, tertiary amines, organometallic compounds, particularly metal carboxylates, and mixtures thereof.

Particularly suitable catalysts include, for example, di-n-butyl tin-bis(mercaptoacetic acid isooctyl ester), dimethyl tin dilaurate, dibutyl tin dilaurate, stannous octoate, lead octoate, triethylene diamine, N-methyl morpholine, and mixtures thereof.

Suitable inorganic filler materials which can be employed herein include, for example, calcium carbonate, alumina trihydrate, talc, bentonite, antimony

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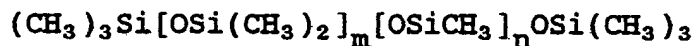
trioxide, kaolin, barytes either as obtained or treated with fatty acids or organotitanates, and mixtures thereof.

5 The inorganic filler is usually employed in quantities of from 25 to 300, preferably from 50 to 200 parts per 100 parts by weight based on the total polyol composition.

10 Suitable silicone surfactants which can be employed herein include the branched organosilicon polyglycol block copolymers having an average molecular weight below 30,000. Such surfactants are described in U.S. Patent 3,518,288 (Haluska).

15 Particularly suitable are DC193 and DC197 commercially available from Dow Corning Corporation; and L5430 commercially available from Union Carbide Corporation.

Particularly suitable silicone surfactants are those represented by the formula



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wherein m has a value of from 6 to 180, preferably from 6 to 60, most preferably from 11 to 20; n has a value of from 3 to 30, preferably from 3 to 10, most preferably from 3 to 5; each x is independently 2, 3 or 4
25 with the proviso that in at least 60, preferably 100 percent, of the $(\text{OC}_{\text{x}}\text{H}_{2\text{x}})$ groups, x has a value of 2; p has a value of from 9 to 30, preferably from 9 to 20, most preferably from 9 to 12; and A is

0048986

selected from $\begin{matrix} \text{O} \\ \parallel \\ -\text{OCR} \end{matrix}$ (acyloxy) especially where
5. R is methyl, $\begin{matrix} \text{O} \\ \parallel \\ -\text{OCOR} \end{matrix}$ (carbonate ester), $\begin{matrix} \text{O} \\ \parallel \\ -\text{OCNHR} \end{matrix}$ (urethane),
-OH (hydroxyl), or -R (alkyl), especially methyl; and D
is an alkylene radical containing from 3 to 18 carbon
atoms, where D is most preferably a propylene radical.
10 These are disclosed by Haluska in U.S. Patent 3,518,288
and Bachura in U.S. Patent 3,706,681.

If desired, the formulations can contain
flame suppressing agents such as, for example, dibromo-
neopentyl glycol, tris(dichloropropyl) phosphate,
15 polyammonium phosphate, beta-chloroethyl phosphonate
ester, chlorophosphate ester, polyvinylidene chloride,
and mixtures thereof.

Suitable substrates to which the frothed
compositions of the present invention are applied
20 include carpet, particularly tufted carpet, paper,
synthetic and natural textile fabrics such as, for
example, nylon, polyester, acrylic, cotton, and wool.

Suitable primary backing materials which can
be employed in the present invention when the substrate
25 is a tufted carpet include those prepared from, for
example, jute, polypropylene, nylon, polyesters, and
polyacrylates.

Suitable yarns which can be tufted or woven
into the primary backing include those prepared from,
30 for example, wool, nylon, polyester, acrylic, cotton,
polypropylene, polyethylene, and blends thereof.

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The following examples are illustrative of the present invention but are not to be construed as to limiting the scope thereof in any manner.

GENERAL PROCEDURE

5 In the examples and comparative runs, a master batch containing a polyol mixture, filler(s) and silicon surfactant was well blended to form a homogeneous creamy dispersion and placed in a run tank maintained at a temperature not less than 45°F (7°C)
10 under a pad of nitrogen. Another run tank containing the polyisocyanate was maintained at a temperature not less than 60°F (16°C). The contents of the run tanks, and catalyst were metered into an Oakes frother where the contents were frothed employing air as the inert
15 gas. The resultant froth was then placed onto the back of the carpet via a hose and doctored to the desired thickness. The froth exit temperature was maintained at a temperature of at least 86°F.(30°C).

20 The carpet containing the polyurethane-forming froth composition was then cured in an oven at a temperature of 250°F (121°C).

25 The carpet onto which the frothed polyurethane-forming compositions were doctored was a flat level loop carpet containing 100 percent continuous filament nylon pile weighing 14 oz/yd² (0.47 Kg/m²) tufted at a gauge of 0.1 inch (3 mm) through an FLW (woven polypropylene needlepunched with nylon or polypropylene fiber) primary backing weighing 4.25 oz/yd² (0.144 Kg/m²) at a stitch rate of 8 per inch and a pile
30 height of 0.125 inch (3.18 mm).

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In the following examples and comparative runs, the properties were determined by the following tests.

- The viscosity of the unfilled urethane forming components is obtained by mixing the components for 3 minute(s) and placing 10 ml. of the resultant mixture in a Brookfield LVF Viscometer, maintaining the temperature of the mixture at $25^{\circ}\text{C} \pm 0.5^{\circ}\text{C}$ and determining the viscosity over the desired period of time.
- 10 The Brookfield LVF Viscometer was employed using a number 4 spindle rotating at 30 rpm.

- The viscosity of a mixture of polyols (A) and (B) or the viscosity of polyol (A) when employed alone is obtained by placing 200 ml of the mixture of polyols (A) and (B) or polyol (A) alone in a Brookfield LVF Viscometer, maintaining the temperature of the material being measured at $25^{\circ}\text{C}, \pm 0.5^{\circ}\text{C}$, and determining the viscosity employing a number 2 spindle rotating at 12 rpm.

- 20 Penetration weight was calculated by subtracting the greige weight (the weight of the carpet sample to which the polyurethane-forming froth composition was applied so as to provide a secondary backing), from the greige weight plus polyurethane-forming materials which had soaked into the tufts and primary
- 25 backing which was the weight of the sample after "skiving" the foam from the back of the carpet sample and brushing with a wire brush to remove as much of the remaining foam as possible.

0048986

The cell count was determined employing a microscope equipped with a Polaroid camera to take photomicrographs of foam and visually count the number of cells along the surface plane in a given area. Each
5 area was ten units by ten units, each unit being 0.005 inches (0.1 mm) (0.0025 inch²)(1.6 mm²). The pictures were taken of the foam surface at or near the center of the foam after splitting the foam for this purpose. The surface plane after splitting was dyed so that
10 individual cells were clearly visible.

The foam density was determined by the following formula:

$$0.08333 \times \frac{\text{coating weight-penetration wt. (oz/yd}^2\text{)}}{\text{Gauge, (inches)}} =$$

15 density, lb/ft³

$$\left[1000 \times \frac{\text{coating weight-penetration wt. (Kg/m}^2\text{)}}{\text{Gauge, (mm)}} = \right.$$

density, (Kg/m³)]

The coating weight was obtained by subtract-
20 ing the greige weight from the total sample weight of the carpet (greige) and foam.

Gauge is the thickness of the foam on the back of the carpet and was measured after the carpet with the foam had been cured.

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The following components were employed in the examples and comparative runs.

The composition of the polyurethane-forming mixtures and the resulting foam properties for
5 Examples 1 to 13 and Comparative Runs A to H are shown in Table I.

Polyol A is a glycerine initiated polyoxypropylene glycol end capped with 15 weight percent ethylene oxide. The resultant primary hydroxyl-containing
10 polyether polyol had an average OH equivalent weight of 1650 and an endcap of an average of 3.75 moles of ethylene per active hydrogen of the polyol.

Polyol B is diethylene glycol having an average OH
15 equivalent weight of 53.

Polyol C is a polyoxyethylene glycol having an average OH equivalent weight of 100.

Polyol D is a glycerine initiated polyoxypropylene glycol having an average OH equivalent weight of
20 87.

Polyol E is a glycerine initiated heteropolyol prepared from a mixture of propylene oxide and ethylene oxide in a ratio of 88/12 weight percent respectively. The resultant secondary hydroxyl-
25 -containing polyether polyol has an average OH equivalent weight of 1270 and an internal block of an average of 2.2 moles of ethylene oxide residues per active hydrogen atom of the polyol.

0048986

Polyol F is dipropylene glycol having an average OH equivalent weight of 67.

Polyol G is a polyoxypropylene glycol having an average OH equivalent weight of 1000.

- 5 Polyol H is a polyoxypropylene triol containing 20 weight percent of a 50/50 styrene/acrylonitrile copolymer and having an average OH equivalent weight of 2000.

- 10 Polyol I is an adduct of a mixture of sucrose and glycerine with propylene oxide having an average functionality of 4.6 and an OH equivalent weight of 130.

Polyol J is a glycerine initiated polyoxypropylene triol having an OH equivalent weight of 240.

- 15 Polyisocyanate A is an 80/20 mixture of 2,4-/2,6--toluene diisocyanate having an average NCO equivalent weight of 87.

- 20 Polyisocyanate B is prepolymer prepared by reacting an excess of polyisocyanate A with a polyol mixture consisting of 64 weight percent of polyol C and 36 weight percent polyol D. The resultant isocyanate-containing prepolymer had an NCO content of 26.7 percent and an average NCO equivalent weight of 157.

- 25 Polyisocyanate C is a carbodiimide modified methylene diphenyldiisocyanate having an NCO equivalent weight of 143 commercially available from Upjohn Company as ISONATE[®] 143L.

[®]Registered Trademark
29,141C-F

0048986

Catalyst A is di-n-butyl tin bis(isooctylmercaptoacetate), commercially available from Witco Chemical Co. as Fomrez®UL-6.

5 Catalyst B is dimethyl tin dilaurate commercially available from Witco Chemical Co. as Fomrez®UL-28.

Filler A is alumina trihydrate commercially available from Kaiser Aluminum Company as H36U.

Filler B is calcium carbonate commercially available from Georgia Marble Company as D-70.

10 Silicone A is a nonhydrolyzable poly(dimethyl siloxane) polyoxyethylene copolymer having a weight average molecular weight of 4000 commercially available from Dow Corning Corporation as DC-197.

15 Silicone B is a 40 percent by weight solution of a linear siloxane-polyoxyalkylene block copolymer having a weight average molecular weight of 100,000 in a hydrocarbon solvent commercially available from Union Carbide Corporation as L 5614.

20 Silicone C is a siloxane-polyoxyalkylene block copolymer having a weight average molecular weight of 40,000, commercially available from Th. Goldschmidt as B 1651. It is a 55 percent weight solution in an aromatic hydrocarbon solvent.

25 Silicone D is a hydrolyzable polysiloxane surfactant having a weight average molecular weight of 4,000, commercially available from Th. Goldschmidt as B 1048.

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Silicone E is a nonhydrolyzable methyl capped siloxane-polyoxyalkylene block copolymer having a weight average molecular weight of 3,000, commercially available from Union Carbide Corporation as L 5340.

- 5 Silicone F is a non-hydrolyzable silicone surfactant having a weight average molecular weight of 15,000 commercially available from Dow Corning Corporation as DC 190.

- 10 Silicone G is a silicone surfactant believed to be diluted in a polyoxyalkylene glycol and is commercially available from Union Carbide Corporation as L 5303.

0048986

TABLE I

	Polyurethane Forming mixture, parts by weight	Compara- tive Run <u>A</u>	Exam- ple <u>1</u>	Exam- ple <u>2</u>	Exam- ple <u>3</u>
5	Comp.(A) poly- ether polyol	A/90	A/90	A/90	A/90
	Comp.(B) polyol	B/10	B/10	B/10	B/10
	Comp.(C) poly- isocyanate	B/46.1	B/46.1	B/46.1	B/46.1
10	Comp.(D), filler	A/40 B/40	A/40 B/40	A/40 B/40	A/40 B/40
	Comp.(E), catalyst	A/0.06	A/0.06	A/0.06	A/0.06
15	Comp.(F), silicone	0	A/0.05	A/0.11	A/0.25
	HEW(A)/ HEW(B)	31/1	31/1	31/1	31/1
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	680 (0.68)	680 (0.68)	680 (0.68)	680 (0.68)
	Viscosity (A),(B),(C),(E) +(F) cps (p.a)	10,600 (10.6)	11,000 (11)	10,600 (10.6)	10,000 (10)
25	at	16 min.	15 min.	16 min.	2 min.

0048986

TABLE I (cont'd)

	Compara- tive Run <u>A</u>	Exam- ple <u>1</u>	Exam- ple <u>2</u>	Exam- ple <u>3</u>
5 <u>Foam properties</u> coating weight, oz/yd ² (Kg/m ²)	45 (1.5)	46 (1.6)	48 (1.6)	54 (1.8)
10 Penetration weight, oz/yd ² (kg/m ²)	16 (0.5)	10 (0.3)	8 (0.3)	6 (0.2)
Gauge, in. (mm)	0.235 (5.97)	0.271 (6.88)	0.294 (7.47)	0.333 (8.46)
15 Cell count, cells per area	26	17	17	35
Foam density, lb/ft ³ (Kg/m ³)	10.3 (165)	11.1 (178)	11.2 (179)	11.9 (191)

0048986

TABLE I (cont'd)

	Polyurethane Forming mixture, <u>parts by weight</u>	Exam- ple <u>4</u>	Exam- ple <u>5</u>	Exam- ple <u>6</u>
5	Comp.(A) poly- ether polyol	A/90	H/90	A/100
	Comp.(B) polyol	B/10	B/10	--
	Comp.(C) poly- isocyanate	C/45.1	B/44	B/11.4
10	Comp.(D), filler	A/55	A/55	A/55
	Comp.(E), catalyst	A/0.01	A/0.07	A/0.06
15	Comp.(F), silicone	A/0.15	A/0.15	A/0.15
	HEW(A)/ HEW(B)	31/1	38/1	--
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	680 (0.68)	1455 (1.455)	*810 (0.81)
	Viscosity (A),(B),(C),(E) +(F) cps (p.a)	10,000 (10)	10,000 (10)	10,000 (10)
25	at	2 min.	8 min.	20 min.

0048986

TABLE I (cont'd)

	Exam- ple <u>4</u>	Exam- ple <u>5</u>	Exam- ple <u>6</u>
5 <u>Foam properties</u>			
coating weight, oz/yd ² (Kg/m ²)	47 (1.6)	44 (1.5)	45 (1.5)
10 Penetration weight, oz/yd ² (kg/m ²)	6 (0.2)	5 (0.2)	11 (0.4)
Gauge, in. (mm)	0.264 (6.71)	0.311 (7.90)	0.231 (5.87)
15 Cell count, cells per area	35	30	19
Foam density, lb/ft ³ (Kg/m ³)	12.8 (205)	10 (160)	12.5 (200)

0048986

TABLE I (cont'd)

	Polyurethane Forming mixture, <u>parts by weight</u>	Exam- ple <u>7</u>	Exam- ple <u>8</u>	Exam- ple <u>9</u>	Exam- ple <u>10</u>
5	Comp.(A) poly- ether polyol	E/88	G/85	I/85	J/85
	Comp.(B) polyol	F/12	F/15	F/15	F/15
	Comp.(C) poly- isocyanate	B/46.1	B/60	B/166	B/108
10	Comp.(D), filler	A/55	A/55	A/55	A/55
	Comp.(E), catalyst	B/0.25	B/0.07	A/0.09	B/0.06
15	Comp.(F), silicone	A/0.10	A/0.25	A/0.15	A/0.45
	HEW(A)/ HEW(B)	19/1	15/1	1.9/1	3.6/1
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	500 (0.5)	245 (0.245)	2260 (2.26)	180 (0.18)
	Viscosity (A), (B), (C), (E) +(F) cps (p.a)	10,400 (10.4)	10,000 (10)	10,000 (10)	10,000 (10)
25	at	13 min.	8 min.	12 min.	9 min.

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TABLE I (cont'd)

	Exam- ple <u>7</u>	Exam- ple <u>8</u>	Exam- ple <u>9</u>	Exam- ple <u>10</u>
5 <u>Foam properties</u>				
coating weight, oz/yd ² (Kg/m ²)	43 (1.5)	46 (1.6)	47 (1.6)	47 (1.6)
10 Penetration weight, oz/yd ² (kg/m ²)	6 (0.2)	10 (0.3)	8 (0.3)	12 (0.4)
Gauge, in. (mm)	0.291 (7.39)	0.264 (6.71)	0.353 (8.97)	0.292 (7.44)
15 Cell count, cells per area	27	17	18	20
Foam density, lb/ft ³ (Kg/m ³)	10.4 (167)	11.2 (179)	9.3 (149)	9.8 (157)

0048986

TABLE I (cont'd)

	Polyurethane Forming mixture, parts by weight	Exam- ple <u>11</u>	Exam- ple <u>12</u>	Exam- ple <u>13</u>
5	Comp.(A) poly- ether polyol	A/90	A/90	A/90
	Comp.(B) polyol	B/10	B/10	B/10
	Comp.(C) poly- isocyanate	B/46.1	B/46.1	B/46.1
10	Comp.(D), filler	A/55	A/55	A/55
	Comp.(E), catalyst	A/0.07	A/0.07	A/0.07
15	Comp.(F), silicone	F/0.15	D/0.15	E/0.15
	HEW(A)/ HEW(B)	31/1	31/1	31/1
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	680 (0.68)	680 (0.68)	680 (0.68)
25	Viscosity (A), (B), (C), (E) +(F) cps (p.a) at	10,500 (10.5) 15 min.	11,000 (11) 16 min.	10,500 (10.5) 15 min.

0048986

TABLE I (cont'd)

	Exam- ple <u>11</u>	Exam- ple <u>12</u>	Exam- ple <u>13</u>
5 <u>Foam properties</u>			
coating weight, oz/yd ² (Kg/m ²)	48 (1.6)	51 (1.7)	47 (1.6)
10 Penetration weight, oz/yd ² (kg/m ²)	8 (0.31)	9 (0.3)	6 (0.2)
Gauge, in. (mm)	0.303 (7.70)	0.303 (7.70)	0.312 (7.92)
15 Cell count, cells per area	19	18	29
Foam density, lb/ft ³ (Kg/m ³)	11 (176)	11.6 (186)	11 (176)

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TABLE I (cont'd)

	Polyurethane Forming mixture, parts by weight	Compar- ative Run B	Compar- ative Run C	Compar- ative Run D	Compar- ative Run E
5	Comp.(A) poly- ether polyol	A/90	A/90	A/90	A/90
	Comp.(B) polyol	B/10	B/10	B/10	B/10
	Comp.(C) poly- isocyanate	B/46.1	B/46.1	B/46.1	B/46.1
10	Comp.(D), filler	A/55	A/55	A/55	A/40 B/40
	Comp.(E), catalyst	A/0.07	A/0.07	A/0.07	A/0.06
15	Comp.(F), silicone	B/0.05	B/0.11	B/0.25	B/0.56
	HEW(A)/ HEW(B)	31/1	31/1	31/1	31/1
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	680 (0.68)	680 (0.68)	680 (0.68)	680 (0.68)
25	Viscosity (A), (B), (C), (E) +(F) cps (p.a) at	11,500 (11.5) 16 min.	12,000 (12) 16 min.	11,500 (11.5) 15 min.	12,800 (12.8) 16 min.

0048986

TABLE I (cont'd)

		Compar- ative <u>Run B</u>	Compar- ative <u>Run C</u>	Compar- ative <u>Run D</u>	Compar- ative <u>Run E</u>
5	<u>Foam properties</u>				
	coating weight, oz/yd ² (Kg/m ²)	47 (1.6)	48 (1.6)	51 (1.7)	50 (1.7)
10	Penetration weight, oz/yd ² (kg/m ²)	14 (0.5)	14 (0.5)	8 (0.3)	5 (0.2)
	Gauge, in. (mm)	0.245 (6.22)	0.224 (5.69)	0.290 (7.37)	0.306 (7.77)
15	Cell count, cells per area	8	2	6	16
	Foam density, lb/ft ³ (Kg/m ³)	11.1 (178)	12.9 (207)	12.3 (197)	12.1 (194)

0048986

TABLE I (cont'd)

	<u>Polyurethane Forming mixture, parts by weight</u>	<u>Compar- ative Run F</u>	<u>Compar- ative Run G</u>	<u>Compar- ative Run H</u>
5	Comp.(A) poly- ether polyol	A/90	A/90	A/90
	Comp.(B) polyol	B/10	B/10	B/10
	Comp.(C) poly- isocyanate	B/46.1	B/46.1	B/46.1
10	Comp.(D), filler	A/55	A/55	A/55
	Comp.(E), catalyst	A/0.07	A/0.07	A/0.07
15	Comp.(F), silicone	C/0.05	C/0.15	C/0.56
	HEW(A)/ HEW(B)	31/1	31/1	31/1
20	Viscosity(A)+(B)) or (A) alone, cps (p.a)	680 (0.68)	680 (0.68)	680 (0.68)
	Viscosity (A),(B),(C),(E) +(F) cps (p.a)	11,500 (11.5)	11,000 (11)	13,000 (13)
25	at	16 min.	16 min.	19 min.

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TABLE I (cont'd)

		Compar- ative <u>Run F</u>	Compar- ative <u>Run G</u>	Compar- ative <u>Run H</u>
5	<u>Foam properties</u> coating weight, oz/yd ² (Kg/m ²)	46 (1.6)	43 (1.5)	47 (1.6)
10	Penetration weight, oz/yd ² (kg/m ²)	15 (0.5)	11 (0.4)	9 (0.3)
	Gauge, in. (mm)	0.248 (6.30)	0.277 (7.04)	0.306 (7.77)
15	Cell count, cells per area	7	13	10
	Foam density, lb/ft ³ (Kg/m ³)	10.4 (167)	9.6 (154)	10.3 (165)

WHAT IS CLAIMED IS:

1. A polyurethane foam composition which is the reaction product of a frothed polyurethane forming mixture comprising:

- (A) a polyethylene polyol having an average of from 2 to 8 hydroxyl groups per molecule and an average hydroxyl equivalent weight of from 100 to 3000;
- (B) a polyol having an average of from 2 to 8 hydroxyl groups per molecule and an average hydroxyl equivalent weight of from 31 to 230, component (B) being optional when component (A) contains an end cap of at least two moles of ethylene oxide per active hydrogen atom or an internal block of at least three moles of ethylene oxide residues per active hydrogen atom, when components (A) and (B) are employed together, they are employed in proportions providing from 50 to 95 parts by weight of component (A) and from 50 to 5 parts by weight of component (B) based upon 100 parts by weight of the total quantity of components

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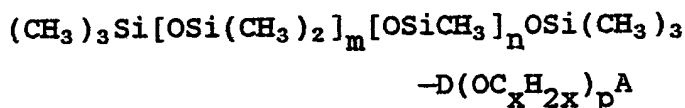
(A) and (B), the hydroxyl equivalent weight of component (A) is at least 1.5 times greater than the hydroxyl equivalent weight of component (B), the mixture of components (A) and (B) or component (A) when employed alone has a Brookfield viscosity employing a number 2 spindle at 12 rpm of at least 100 centipoise (0.1 Pascal-second) at 25°C;

- (C) an organic polyisocyanate or polyisothiocyanate having an average of from 2 to 4 NCX groups per molecule, each X being independently oxygen or sulfur, components (A), (B) and (C) being employed in quantities so as to provide an NCX ratio of from 0.85:1 to 2:1;
- (D) at least one inorganic filler;
- (E) at least one catalyst for reacting components (A) and (B) with component (C), component (E) being employed in a quantity such that an unfrothed mixture of components (A), (B), (C), (E) and (F) obtains a Brookfield viscosity employing a number 4 spindle at 30 rpm greater than 10,000 centipoise (10 Pascal-seconds) at 25°C in less than 40 minutes; and
- (F) a silicone surfactant, characterized in that component (F) is a poly(dimethylsiloxane) polyoxyalkylene branch copolymer having an average weight of less than 30,000, wherein at least 60 percent by weight of the polyoxyalkylene groups are oxyethylene groups, the dimethylsiloxane content of the copolymer being from 15 to 40 percent by weight, and at least 40 percent of the weight of the copolymer is derived from the oxyethylene groups,

0048986

component (F) being present in a quantity from 0.02 to less than 0.5 parts by weight per 100 parts by weight of the combined weights of components (A) and (B).

2. The foam composition of Claim 1 characterized in that component (F) is represented by the formula



wherein m has a value of from 6 to 180, n has a value of from 3 to 30, each x is independently 2, 3 or 4 with the proviso that in at least 60 percent of the $(\text{OC}_x\text{H}_{2x})$ groups, x has a value of 2; p has a value of from 9 to 30; A is selected from

$\begin{matrix} \text{O} \\ \parallel \\ -\text{OCR} \end{matrix}$ (acyloxy) especially where

R is methyl, $\begin{matrix} \text{O} \\ \parallel \\ -\text{OCOR} \end{matrix}$ (carbonate ester), $\begin{matrix} \text{O} \\ \parallel \\ -\text{OCNHR} \end{matrix}$ (urethane), -OH (hydroxyl), or -R (alkyl); and D is an alkylene radical containing from 3 to 18 carbon atoms.

3. The foam composition of Claim 2 characterized in that m has a value of from 6 to 60; n has a value of from 3 to 10; each x has a value of 2; p has a value of from 9 to 20; A is acetate, urethane, hydroxyl or methyl.

4. The foam composition of Claim 3 characterized in that m has a value of from 11 to 20; n has a value of from 3 to 5; p has a value of from 9 to 12 and D is the propylene radical.

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5. A substrate having a polyurethane foam backing containing an inorganic filler or mixture of such filler characterized in that the polyurethane foam is the composition of any of Claims 1, 2 or 3.

10.10.81

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PATENTANWÄLTE

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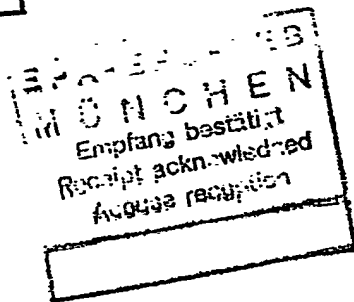
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The Dow Chemical Company

(EU 218 - Case 29,141C-F)



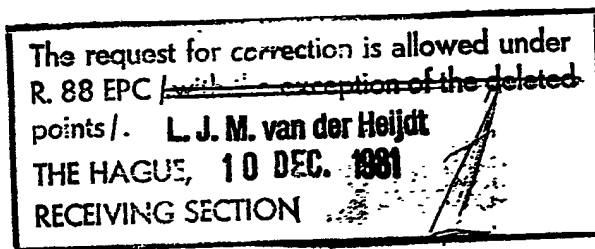
Enclosed please find description pages 9, 10 and 32 where obvious typing errors have been corrected. It is requested to replace the version, presently on file, by the corrected one.



Encls.

Amended descript.
pages 9,10,32

Dipl.-Chem. B. Huber



EUROPEAN PATENT OFFICE
ERHARDTSTR. 27
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selected from $\begin{array}{c} \text{O} \\ \parallel \\ -\text{O}-\text{C}-\text{R} \end{array}$ (acyloxy) especially where
5 R is methyl, $\begin{array}{c} \text{O} \\ \parallel \\ -\text{O}-\text{C}-\text{O}-\text{R} \end{array}$ (carbonate ester), $\begin{array}{c} \text{O} \\ \parallel \\ -\text{O}-\text{C}-\text{N}-\text{H}-\text{R} \end{array}$ (urethane),
-OH (hydroxyl), or -R (alkyl), especially methyl; and D
is an alkylene radical containing from 3 to 18 carbon
atoms, where D is most preferably a propylene radical.
10 These are disclosed by Haluska in U.S. Patent 3,518,288
and Bachura in U.S. Patent 3,706,681.

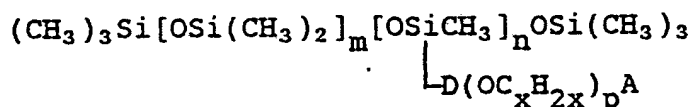
If desired, the formulations can contain
flame suppressing agents such as, for example, dibromo-
neopentyl glycol, triis(dichloropropyl) phosphate,
15 polyammonium phosphate, beta-chloroethyl phosphonate
ester, chlorophosphate ester, polyvinylidene chloride,
and mixtures thereof.

Suitable substrates to which the frothed
compositions of the present invention are applied
20 include carpet, particularly tufted carpet, paper,
synthetic and natural textile fabrics such as, for
example, nylon, polyester, acrylic, cotton, and wool.

Suitable primary backing materials which can
be employed in the present invention when the substrate
25 is a tufted carpet include those prepared from, for
example, jute, polypropylene, nylon, polyesters, and
polyacrylates.

Suitable yarns which can be tufted or woven
into the primary backing include those prepared from,
30 for example, wool, nylon, polyester, acrylic, cotton,
polypropylene, polyethylene, and blends thereof.

2. The foam composition of Claim 1 characterized in that component (F) is represented by the formula



0
"
-OCR (acyloxy) especially where

3. The foam composition of Claim 2 characterized in that m has a value of from 6 to 60; n has a value of from 3 to 10; each x has a value of 2; p has a value of from 9 to 20; A is acetate, urethane, hydroxyl or methyl.

29,141C-F



DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.7)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
	<p>DE - A1 - 2 748 085 (MOBAY CHEMICAL CORP.)</p> <p>* claim 3; page 8, lines 5 to 13; page 13, line 20 to page 14, line 13; page 15, lines 6 to 15, page 18, lines 10 to 17 *</p> <p>& GB - A - 1 573 778</p>	1	C 08 L 75/04 B 32 B 27/40 C 08 G 18/14 C 08 G 18/61 C 08 G 77/46 C 08 J 9/30 D 06 N 3/14 D 06 N 7/00
	<p>GB - A - 1 383 921 (TEXTILE RUBBER AND CHEMICAL COMPANY INC.)</p> <p>* claims 15 to 17; page 3, lines 27 to 42; page 5, lines 38 to 59 *</p>	1,5	TECHNICAL FIELDS SEARCHED (Int. Cl.7)
	<p>GB - A - 1 306 372 (UNION CARBIDE CORP.)</p> <p>* claims 1, 13, 15, 16, 20 *</p>	1,5	B 32 B 27/40 C 08 G 18/00 C 08 G 77/46 C 08 J 9/30 C 08 L 75/04 D 06 N 3/00 D 06 N 7/00
D	<p>US - A - 3 862 879 (B.G. BARRON et al.)</p> <p>* column 1, lines 34 to 65; column 3, lines 35 to 42; column 4, lines 59 to 66 *</p>	1	CATEGORY OF CITED DOCUMENTS
	<p>US - A - 3 706 681 (R.A. BACHURA)</p> <p>* column 1, lines 13 to 19; column 2, lines 20 to 61 *</p>	2,3	X: particularly relevant A: technological background O: non-written disclosure P: intermediate document T: theory or principle underlying the invention E: conflicting application D: document cited in the application L: citation for other reasons
D	<p>US - A - 3 518 288 (L.A. HALUSKA)</p> <p>* claim 5; column 2, line 15 to column 4, line 24 *</p>	2-4	
			&: member of the same patent family, corresponding document
The present search report has been drawn up for all claims			
Place of search	Date of completion of the search	Examiner	
Berlin	15-12-1981	IDEZ	



European Patent
Office

EUROPEAN SEARCH REPORT

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Application number

EP 81 10 7699.1

- page 2 -

D CUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.7)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
A	<p>BE - A - 676 149 (UNION CARBIDE CORP.)</p> <p>* page 2, line 3 to page 5, line 8 *</p>		
			TECHNICAL FIELDS SEARCHED (Int. Cl.7)